

--	--	--	--	--	--	--	--	--	--



**DHANALAKSHMI SRINIVASAN COLLEGE  
OF ARTS & SCIENCE FOR WOMEN  
(AUTONOMOUS)**

(For Candidates admitted from 2019-2020 onwards)

**PG DEGREE EXAMINATIONS APRIL - 2021**

**M.Sc., - CHEMISTRY**

**PHYSICAL CHEMISTRY- II**

**Time: 3 Hrs**

**Max.Marks: 75**

**PART - A**

**CHOOSE THE CORRECT ANSWER**

**(10X1=10)**

- The ground state energy of helium atom according to variation the theorem is
  - 77.48 eV
  - 110.28 eV
  - 2.32 au
  - 3.12 au
- The term symbols for 2p1 electron are
  - 2p1, 2p1/2
  - 2p1/2, 2p3/2
  - 2p3/2, 2p1
  - 2p1/2, 2p2
- The transference number of an ion
  - is always positive
  - is always negative
  - can be positive as well as negative
  - is always zero
- Variation of molar conductivity with concentration of strong electrolyte is given by Huckel-Onsager equation expressed as
  - $\Lambda_M = \Lambda_\alpha - b\sqrt{c}$
  - $\Lambda_\alpha = \Lambda_M - b\sqrt{c}$
  - $\Lambda_M = b\sqrt{c} - \Lambda_\alpha$
  - None of these
- Which of the following is an example of homogeneous catalysis?
  - Enzyme catalysis
  - Hardening of animal and vegetable oils
  - Herber's process
  - Cracking of heavy oils for a synthesis of gasoline
- The rate determining step of Michaelis-Menten kinetics is
  - The Complex formation step
  - The product formation step
  - The complex dissociation step to produce products
  - None of the above
- For an idea gas the free energy of mixing is given by
  - $\Delta G_{\text{mix}} = \sum \chi_i \ln \chi_i / RT$
  - $\Delta G_{\text{mix}} = \sum \chi_i \ln \chi_i - RT$
  - $\Delta G_{\text{mix}} = \sum \chi_i \ln \chi_i + RT$
  - $\Delta G_{\text{mix}} = RT \sum \chi_i \ln \chi_i$
- The total vapour pressure of an ideal gas is given by
  - $P = x_i / p_i$
  - $P = x_i + p_i$
  - $P = \sum p_i$
  - $P = x_i - p_i$
- What is Miller indicis of a plane which makes intercepts 2a, 3b, 2c?
  - 2 : 3 : 2
  - 3 : 2 : 3
  - 2 : 2 : 3
  - 3 : 2 : 2
- The total number of space group in a crystal is
  - 7
  - 14
  - 32
  - 230

**PART- B**

**ANSWER ALL THE QUESTIONS**

**(5X7=35)**

11. a) Write short notes on Slater type orbitals.

**(OR)**

b) Discuss Hellman-Feynman theorem.

12. a) What is Debye-Huckel theory.

**(OR)**

b) Discuss about Evans diagrams.

13. a) Discuss ultrasonic absorption techniques.

**(OR)**

b) Write notes on Langmuir-Hinshelwood mechanism.

14. a) Write short notes on Maxwell relations.

**(OR)**

b) Discuss the free energy change of mixing.

15. a) Explain the Bragg's law.

**(OR)**

b) Write short notes on phase problem.

**PART-C**

**ANSWER ANY THREE QUESTIONS**

**(3X10=30)**

16. (i) Explain the LCAO-MO.

(ii) Write notes on selection rule.

17. Derive Butler-Volmer equation.

18. (i) What is meant by homogeneous catalysis.

(ii) Explain the Michaelis-Menten kinetics

19. Derive Clausius-Clapeyron equation

20. Write short notes on:

(i) Miller Indices

(ii) Laue equation

(iii) Fourier Transform

(iv) space groups